

Half-Metallic Ferromagnetism and Surface Functionalization-Induced Metal–Insulator Transition in Graphene-like Two-Dimensional Cr₂C Crystals

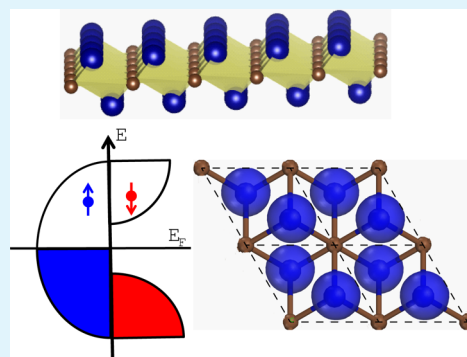
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S Supporting Information

ABSTRACT: Graphene-like two-dimensional materials have garnered tremendous interest as emerging device materials for nanoelectronics due to their remarkable properties. However, their applications in spintronics have been limited by the lack of intrinsic magnetism. Here, using hybrid density functional theory, we predict ferromagnetic behavior in a graphene-like two-dimensional Cr₂C crystal that belongs to the MXenes family. The ferromagnetism, arising from the itinerant Cr d electrons, introduces intrinsic half-metallicity in Cr₂C MXene, with the half-metallic gap as large as 2.85 eV. We also demonstrate a ferromagnetic–antiferromagnetic transition accompanied by a metal to insulator transition in Cr₂C, caused by surface functionalization with F, OH, H, or Cl groups. Moreover, the energy gap of the antiferromagnetic insulating state is controllable by changing the type of functional groups. We further point out that the localization of Cr d electrons induced by the surface functionalization is responsible for the ferromagnetic–antiferromagnetic and metal to insulator transitions. Our results highlight a new promising material with tunable magnetic and electronic properties toward nanoscale spintronics and electronics applications.

KEYWORDS: MXene, half-metallicity, ferromagnetism, functionalization, spintronics



I. INTRODUCTION

The discovery of graphene has spurred a surge in the study of two-dimensional (2D) materials during the past decade. Due to the dimension and size reduction and the resultant quantum effects, 2D crystals have shown many intriguing properties that are not found in their bulk counterparts,^{1,2} which are hence considered as cornerstones of the future nanoscale electronics and spintronics.^{3,4} Although more and more 2D crystals have been produced experimentally, the vast majority of them are nonmagnetic. Thus, the pursuit of tunable magnetism in 2D crystals has been a persistent goal for a long time.^{5–8} In graphene and monolayer transition metal dichalcogenides, several ideas have been proposed to induce the magnetic moments and orderings, such as depositing magnetic adatoms on the surface^{9–11} and introducing specific defects^{5,6} or edges.^{12–14} However, the experimental realization remains challenging: for example, the clustering of the adatoms is always inevitable,¹⁵ and the edge morphology and defect type are not well controllable.¹⁶ Another important property highly desirable for 2D materials, especially for their electronics applications, is the bandgap tunability, which would allow great flexibility in design and optimization of 2D-materials-based devices such as p–n junctions, transistors, photodiodes, and lasers. Though various methods for opening/tuning the gap of 2D material have been developed that include size quantization,¹⁷ chemical functionalization,¹⁸ strain engineering,¹⁹

electric field control,²⁰ and so on, the controllable modulation of the gap is somewhat tough and bandgap engineering is still one of the hottest subjects in the field of 2D materials.

Recently, a new family of 2D early transition metal carbides and/or nitrides labeled “MXenes” has been synthesized by the extraction of “A” layers from MAX phases, which are layered ternary carbides and carbonitrides with a general formula M_{n+1}AX_n (n = 1, 2, 3), where M is an early transition metal, A is mainly a group 13 or 14 element, and X is carbon and/or nitrogen.²¹ The structure of the MAX phase is comprised of stacks of “M_{n+1}X_n” layers interleaved with “A” layers.^{22,23} The A layers can be chemically etched without disrupting M–X bonds, and then the resulting weakly bonded “M_{n+1}X_n” layers can be readily exfoliated into MXenes.²¹ To date, the MXenes family has included Ti₃C₂, Ti₂C, V₂C, Nb₂C, Ta₄C₃, Nb₄C₃, Ti₃CN, (Ti_{0.5}Nb_{0.5})₂C, and (V_{0.5}Cr_{0.5})₃C₂.²¹ Moreover, because there are over 70 known MAX phases, many more MXenes are expected to be synthesized experimentally. Since their discovery, MXenes have attracted increasing attention,^{24–28} showing tremendous promise for applications in electrode materials, sensors, catalysis, and electrochemical energy storage.^{29–33}

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In this work, using hybrid density functional theory calculations, we predict a half-metallic ferromagnet in the family of MXenes, Cr₂C MXene, which is expected to be produced by selectively etching Al atoms from the MAX phase Cr₂AlC, similar to Ti₂C and V₂C from Ti₂AlC and V₂AlC, respectively. The ferromagnetism in Cr₂C MXene is arising from the itinerant Cr d electrons which are 100% spin-polarized around the Fermi surface. And the half-metallic gap is as large as 2.85 eV. We also find that, after surface functionalization with F, OH, H, or Cl groups, the Cr₂C MXene will undergo a metal to insulator transition which is accompanied by a ferromagnetic (FM)–antiferromagnetic (AFM) transition. Moreover, the energy gap in the AFM state is controllable by changing the type of surface functionalization. We further point out that the localization of Cr d electrons induced by the functionalization is responsible for the FM–AFM and metal to insulator transitions. Our results highlight a promising material in the MXenes family, with tunable magnetic and electronic properties, extending the potential applications of MXenes into spintronics and electronics.

II. MODELS AND METHODS

Our calculations are performed in the framework of density functional theory as implemented in the Vienna *ab initio* simulation package (VASP).³⁴ The projector augmented wave (PAW) potential is used with the plane-wave cutoff energy set as 400 eV. The Heyd–Scuseria–Ernzerhof (HSE) screened Coulombic hybrid density functional³⁵ is used to describe the electron correlation effects in the structural optimizations as well as in the static calculations. The system is modeled by a pristine or functionalized Cr₂C MXene single layer and a vacuum region more than 15 Å to avoid interaction between neighboring slabs. The Brillouin zone is integrated with a 20 × 20 × 1 *k* mesh. All of the structures are fully relaxed until the remaining force acting on each atom is less than 0.01 eV/Å.

III. RESULTS AND DISCUSSION

Figure 1a shows the lattice structure of Cr₂C MXene. It is comprised of triple layers where a C atomic layer is sandwiched between two Cr layers. From its top view, we can see the Cr

atoms are arranged in hexagonal structure, where the Cr atoms in two different layers form two triangular sublattices. It is found that Cr₂C MXene energetically favors the spin-polarized ground state, and the energy difference between the spin-polarized and spin-unpolarized states is large, which is about 2.53 eV per unit cell containing two Cr atoms and one C atom. In the spin-polarized ground state, the equilibrium lattice constant is calculated to be 3.14 Å (see Figure 1b), with the thickness of the Cr₂C triple layers (*h*) and the Cr–C bond length being 2.10 and 2.12 Å, respectively. The structural stability of Cr₂C MXene is also verified by the phonon spectra calculations where no negative frequency phonon is found at any wave vector [see the Supporting Information]. The total magnetic moment per unit cell is calculated to be 8 μ_B, mainly contributed to by the two Cr atoms.

In order to find the preferred magnetic coupling between the Cr atoms, we use a 2 × 1 supercell that contains four Cr atoms. Depending on the initial conditions of the self-consistent calculations, we obtain three stable magnetic structures: FM (Figure 1c), AFM1 (Figure 1d), and AFM2 (Figure 1e). (Note that the AFM1 and AFM2 structures shall not be considered as realistic AFM orderings. They are studied only for obtaining the magnetic interaction.) Taking the energy of the FM structure as the reference, the relative energies of the AFM1 and AFM2 structures are 0.36 and 1.06 eV, respectively. The exchange interaction can be conveniently studied by mapping the total energies of the systems with different magnetic orderings to a Heisenberg model with the nearest- and next-nearest-neighbor couplings *J*₁ and *J*₂:

$$H = - \sum_{i,j} J_1 S_i \cdot S_j - \sum_{k,l} J_2 S_k \cdot S_l \quad (1)$$

where *S*_{*i*} is the net spin at the Cr sites *i* and (*i*, *j*) and (*k*, *l*) are the nearest site pairs and next-nearest site pairs, respectively. Based on this model, the energy difference between the FM and the AFM1 phases is $E^{\text{AFM1}} - E^{\text{FM}} = 12J_1 S^2$, and the energy difference between the FM and AFM2 phases is $E^{\text{AFM2}} - E^{\text{FM}} = (4J_1 + 16J_2) S^2$. Then the exchange coupling parameters are calculated to be *J*₁ = 7.4 meV and *J*₂ = 14.7 meV. Their positive values clearly indicate that Cr₂C MXene prefers FM coupling. Moreover, based on *ab initio* molecular dynamics simulations, we further find the magnetic state of Cr₂C could survive at room temperature (see the Supporting Information).

Having studied the magnetic ground state of Cr₂C MXene, we now turn to its electronic structure. As shown in Figure 2a, the most striking property of this band structure is the half-metallicity, i.e., a metallic character found for the majority-spin electrons while an insulating nature is found for the minority-spin electrons. Thus, the charge transport is dominated by the majority-spin electrons, and the electrical current in such a system should be completely spin-polarized. The half-metallic gap, defined as the difference between the Fermi level and topmost occupied minority-spin band,³⁶ is as large as 2.85 eV, suggesting that the 100% spin-filter efficiency can be maintained in a broad bias range. It is stressed that the half-metallicity in Cr₂C MXene is completely intrinsic, in contrast to other low-dimensional materials, such as graphene or boron nitride nanoribbon,^{12,36–38} C/BN heterostructures,^{39,40} and MoS₂⁴¹ and MnPSe₃ nanosheets,⁴² where the presence of half-metallicity needs either a strong external electric field or the carefully selective doping. Such an intriguing physical property, half-metallic ferromagnetism, makes Cr₂C MXene an attractive candidate for nanoscale spintronics applications. It is worth

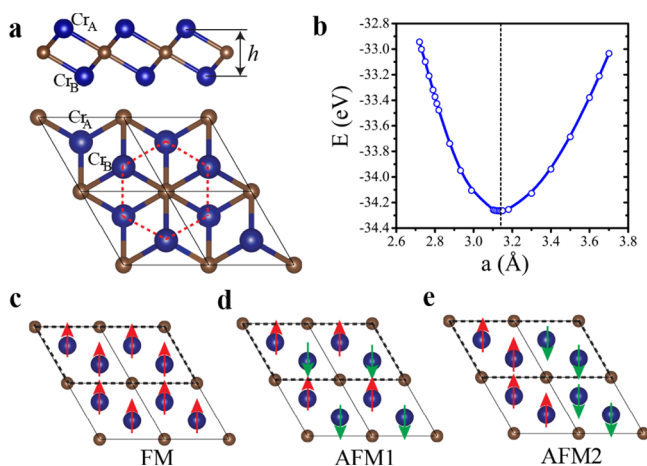


Figure 1. (a) Side and top views of Cr₂C MXene lattice. Blue and brown balls represent Cr and C atoms, respectively. The unit cell of Cr₂C MXene contains two Cr atoms (Cr_A and Cr_B) and one C atom. (b) Variation of total energy of Cr₂C MXene with the lattice constant. (c–e) Schematics of three different magnetic structures: FM, AFM1, and AFM2. The dashed line outlines the 2 × 1 supercell used for total energy calculations.

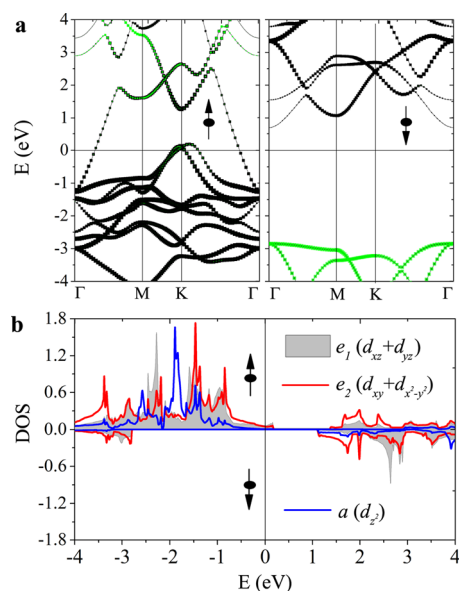


Figure 2. (a) Band structure for Cr_2C MXene. The black and green squares represent the weights of the Cr d and C p orbitals, respectively. (b), Partial density of states of Cr d orbitals. The Fermi level in a and b is set to zero.

mentioning that Cr_2C MXene is the first reported half-metallic ferromagnet among the MXenes family. Even though its ferromagnetic property was concluded by simply comparing the energies of the ferromagnetic and nonmagnetic states in a recent work by Khazaei et al., the half-metallicity of 2D Cr_2C was not discovered therein, probably due to the lack of extensive study on electronic structures.⁴³

Another remarkable feature of the band structure is the strongly dispersing bands crossing the Fermi level (E_F), which provides the clues to the basic mechanism for the ferromagnetic metal behavior. The atomic and orbital resolved characters of the energy bands show a predominant contribution of Cr d orbitals to the dispersing bands that cross the E_F (see Figure 2a). This means the Cr d electronic states around the E_F are itinerant. According to the Stoner theory, the itinerant d electrons would favor ferromagnetism.⁴⁴ In Cr_2C MXene, each Cr^{2+} ion has a closed shell Ar core and four additional 3d electrons. Under the C_{3v} symmetry of the crystal field, the 3d orbitals of Cr split into a single a (d_z^2) orbital and two 2-fold degenerate e_1 ($d_{xz} + d_{yz}$) and e_2 ($d_{xy} + d_{x^2 - y^2}$) orbitals. Because of the large exchange splitting of the Cr 3d orbitals, the four 3d electrons only occupy the majority-spin channel, resulting in an energy gap between the occupied C p orbitals and the empty Cr d orbitals in the minority-spin channel straddling the Fermi energy (see Figure 2a). Then why is the majority-spin channel metallic? Here we point out that this is associated with the nonlocalized nature of the d orbitals. Figure 2b displays the projected density of states (PDOS) of a, e_1 , and e_2 orbitals. It is clearly seen that the a, e_1 , and e_2 orbitals are not localized, showing wide peaks and overlapping with each other. Consequently, in the majority-spin channel, all of them are fractionally occupied, resulting in the appearance of the metallicity.

At present, the main synthesis method for MXenes is etching Al from MAX phases in the hydrofluoric acid or a solution of lithium fluoride and hydrochloric acid,^{21,25} making the termination of MXenes surfaces by OH or F groups possible.²¹

It is widely known that the surface functionalization is one of the main methods to manipulate the physical and chemical properties of 2D materials.^{18,45} Thus, we have also studied the functionalized Cr_2C MXene, i.e., Cr_2C MXene with surface fully terminated by possible functional groups, denoted by a general formula of Cr_2CT_2 , where T stands for surface terminations (F, OH, H, or Cl).

Parts a and b of Figure 3 show the lattice structure of Cr_2CF_2 . It is determined by considering and comparing all

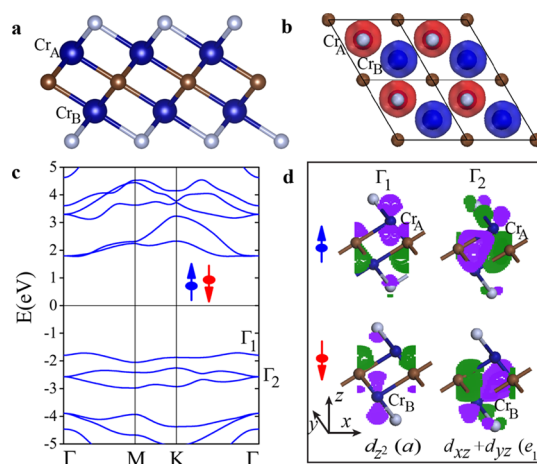


Figure 3. (a, b) Side and top views of optimized structure of Cr_2CF_2 . Blue, brown, and gray balls denote the Cr, C, and F atoms, respectively. The spin density ($\rho_\uparrow - \rho_\downarrow$) map of Cr_2CF_2 is also shown in b. The blue and red colors represent opposite spin orientations: spin up and spin down. (c) Band structure of Cr_2CF_2 under the AFM ground state. The blue lines are for the majority-spin and the red lines are for the minority-spin, where the blue lines completely overlap with the red lines. (d) The wave function for the three highest valence states at the Γ point (Γ_1 and doubly degenerate Γ_2) in the majority- and minority-spin channel. The Γ_1 and Γ_2 states are mainly contributed to by the Cr d_z^2 (a) and $d_{xz} + d_{yz}$ (e_1) orbitals, respectively. The small contribution from C atoms is due to the weak covalence of the Cr_2CF_2 system.

possible adsorption configurations of F atoms on the surface of Cr_2C MXene. One can see that on both sides of Cr_2C , F atoms are located above the hollow sites of three neighboring C atoms, pointing to the Cr atoms in the nonadjacent Cr layer. We stress that the fluorination will not break the structural stability of the system: by calculating the phonon spectra of Cr_2CF_2 , we find that all of the phonon frequencies are positive (see the Supporting Information). Moreover, after fluorination, the magnetic moments on the Cr atoms do not vanish. In order to explore the magnetic ground state of Cr_2CF_2 , we use a 2×2 supercell and consider the nonmagnetic (NM) structure, the FM structure, and all of the possible AFM magnetic structures. We find the ground state is AFM where all of the nearest Cr neighbors have antiparallel spins (corresponding to the AFM1 structure in Figure 1e). This AFM state is more stable than the FM state by an energy difference of 0.60 eV per unit cell containing two Cr atoms. Figure 3b shows the spin density map of Cr_2CF_2 at the AFM ground state, where one can clearly see that the magnetic moments are mainly contributed to by the Cr atoms and the nearest Cr atoms antiferromagnetically couple with each other. The local magnetic moment on each Cr atom is about $3 \mu_B$.

The band structure of Cr_2CF_2 is shown in Figure 3c, where semiconducting behaviors are found for both the majority- and

minority-spin electrons. The band gap is indirect, about 3.49 eV, much larger than the monolayer MoS₂ (1.8 eV)⁴⁶ and phosphorene (2.1 eV)⁴⁷ but smaller than the *h*-BN single layer (5.97 eV)⁴⁸ and graphane (5.4 eV).¹⁸ It is also noted that the three highest valence bands, mainly contributed to by Cr *d* orbitals, are nearly dispersionless with very narrow bandwidth. This distinctly indicates the localized character of *d* electrons, which we will show is a dominant reason for the insulating behavior of Cr₂CF₂ in the following. The Cr atom in Cr₂CF₂ is under the *D*_{3d} symmetry that split 3*d* orbitals into a single *a* (*d*_{z²}) orbital and two doubly degenerate *e*₁ (*d*_{xz} + *d*_{yz}) and *e*₂ (*d*_{xy} + *d*_{x²-y²}) orbitals. In its 3+ valence state, each Cr ion has three 3*d* electrons. Due to the large exchange splitting of Cr *d* orbitals, the three 3*d* electrons only occupy one spin channel. Moreover, because of the localized character of the *d* orbitals, the *e*₁ and *a* orbitals in one spin channel are completely occupied in sequence, while the *e*₂ orbitals are unoccupied, resulting in the formation of semiconductivity. The occupation of *e*₁ and *a* orbitals could be verified by the wave function of the highest valence states at the Γ point. As shown in Figure 3d, in the majority channel, at the Γ point, the highest occupied states (Γ_1) are mainly contributed to by the Cr_A *d*_{z²} (*a*) orbital, and the second highest states (Γ_2), which are 2-fold degenerate, are mainly contributed to by the Cr_A *d*_{xz} + *d*_{yz} (*e*₁) orbitals. Correspondingly, in the minority-spin channel, Γ_1 and Γ_2 are mainly contributed to by Cr_B *d*_{z²} (*a*) and *d*_{xz} + *d*_{yz} (*e*₁) orbitals, respectively.

After knowing the electron configuration of Cr³⁺ ion in Cr₂CF₂, it is not surprising that Cr³⁺ ions couple antiferromagnetically. This is because the *e*₁ and *a* orbitals of a Cr³⁺ ion in one spin channel are fully occupied; virtual hopping is allowed in the AFM arrangement but not allowed in the FM configuration (see Figure 4a), resulting in a lower energy AFM state.

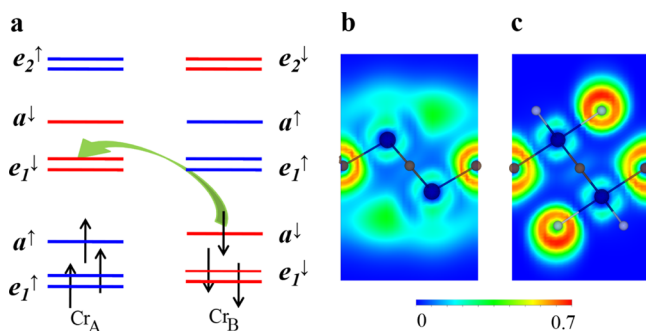


Figure 4. (a) Schematic of the exchange mechanism for Cr₂CF₂. The *e*₁ and *a* orbitals of a Cr³⁺ ion are occupied in one spin channel, and thus virtual hopping results in the AFM arrangement. (b, c) ELF maps of Cr₂C and Cr₂CF₂, respectively.

So far, we have shown that, from Cr₂C to Cr₂CF₂, fluorination induces an interesting metal to insulator transition which is accompanied by a FM–AFM transition. We have also pointed out that the completely different electronic and magnetic properties of Cr₂C and Cr₂CF₂ are closely related to different characteristics of their *d* orbitals: the former has itinerant *d* orbitals, while the latter has localized *d* orbitals. This could be qualitatively reflected by their electron-localization function (ELF). As shown in Figure 4b,c, a delocalized ELF feature for Cr₂C is in contrast to a localized ELF feature for Cr₂CF₂.

Other functionalized MXenes, Cr₂C(OH)₂, Cr₂CH₂, and Cr₂CCl₂, show lattice structures and antiferromagnetic ground states similar to those of Cr₂CF₂. All of them are semiconductors, with the sizes of energy gaps being 1.43 eV for Cr₂C(OH)₂, 1.76 eV for Cr₂CH₂, 2.56 eV for Cr₂CCl₂, and 3.49 eV for Cr₂CF₂ (see Figure 5). That is to say, by choosing

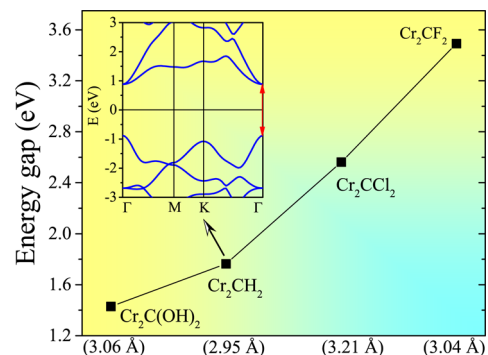


Figure 5. Energy gap of Cr₂CT₂ (T = F, OH, H, or Cl) MXenes. The numbers in the brackets under the *x* axis are the lattice constants for corresponding Cr₂CT₂ MXenes. Inset: Band structure of Cr₂CH₂ with the direct gap at the Γ point highlighted by a red arrow. The Fermi level is set to zero.

different surface functional groups, one can change the metallic Cr₂C MXene into a semiconductor and tune the band gap to different regimes. Most remarkably, Cr₂CH₂ has a direct band gap, with the conduction band minimum and the valence band maximum both located at the Γ point, as shown in the inset of Figure 5. The direct gap that facilitates efficient light emission, makes Cr₂CH₂ an attractive material for the applications in optoelectronic nanodevices. Earlier works have successively found six types of terminated MXenes which are semiconductors: Ti₂CO₂, Zr₂CO₂, Hf₂CO₂, Sc₂CO₂, Sc₂C(OH)₂, and Sc₂CF₂.^{43,49,50} The finding of semiconducting Cr₂CT₂ (T = F, OH, H, or Cl), further expands the semiconducting MXenes family. Finally, we should stress the significant difference between Cr₂CT₂ and the six other types of semiconducting MXenes mentioned earlier: the former are magnetic, while the latter are nonmagnetic.

IV. CONCLUSIONS

In conclusion, we have studied the electronic and magnetic properties of Cr₂C MXene based on hybrid density functional theory calculations. We show that Cr₂C MXene exhibits half-metallic ferromagnetic behavior with a large half-metallic gap of 2.85 eV. The ferromagnetism is arising from the itinerant Cr *d* electrons which are 100% spin-polarized around the Fermi surface. We also demonstrate a FM–AFM transition in Cr₂C MXene caused by surface functionalization with F, H, OH, or Cl groups, which is accompanied by a metal to insulator transition. It is noted that the size of the band gap in the AFM insulating state is controllable by selecting different functional groups. And hydrogenation can open a direct band gap in Cr₂C MXene. We further point out that the functionalization induced localization of Cr *d* electrons is the underlying mechanism for the FM–AFM and metal to insulator transitions. Our results demonstrate the potential for the utilization of Cr₂C MXene in innovative spintronics and electronics devices.

■ ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acsami.5b05401.

Phonon spectra and molecular dynamics simulations for the Cr₂C MXene (PDF)

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Notes

The authors declare no competing financial interest.

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